

# Seaforth Mineral & Ore Fluorspar (Acid Grade)

# **Safety Data Sheet**

Distributed by: Laguna Clay Company 14400 Lomitas Ave City of Industry, CA 91746 1-800-4Laguna info@lagunaclay.com www.lagunaclay.com

#### 1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

**Product Name:** Fluorite, Fluorspar

**Recommended Use:** Flux in ferrous metallurgy, Glass & Hydrofluoric Acid Production

Seaforth Mineral & Ore Co., Inc. **Manufactured For:** 

> 3690 Orange Place, Suite 495 Cleveland, Ohio, 44122 Phone: (216) 292-5820

**Emergency Telephone:** 800-292-9022 Distributed by: Laguna Clay Company 14400 Lomitas Ave

City of Industry, CA 91746

1-800-4Laguna info@lagunaclay.com www.lagunaclay.com

#### 2. HAZARDS IDENTIFICATION

Classification and Labeling: (Regulation (EC) No 1272/2008) CLP/GHS)

**Hazard Statements:** Irritation (Eye) (2): H319-Causes serious eye irritation. Irritation (Skin) (2):

H315-Causes skin irritation. STOT SE (3): H335-May cause respiratory

irritation.

Signal Word: Warning Hazard Pictogram: Irritant.



**Precautionary Statements:** Wash (Hands, Face, contaminated skin) thoroughly after handling. Do not eat, drink or smoke when using this product. Use personal protective equipment as required. May be Harmful if swallowed or inhaled. May cause irritation to skin, eyes and respiratory tract.

**In Eyes**: Rinse Cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

If on Skin: Wash with plenty of soap and water. If skin irritation occurs, get medical advice. Wash contaminated clothing before reuse.

**If inhaled:** Remove victim to fresh air and keep at rest in a position comfortable for breathing.

**If Swallowed:** Call a poison center or doctor/physician. Rinse mouth.

Containers: Dispose of contents/containers in full compliance with Federal, State and local regulations. Other Hazards: Reactive with Acids. Release of hydrogen fluoride (toxic and corrosive gas) under specific moisture and very high temperature conditions. Dust: Possibility of mechanical; irritation of eyes and skin, mucous membranes, upper respiratory tract, lungs.

WHMIS Classification (Canada): Not regulated.

#### **Not Regulated by DOT (USA-HCS)**

**Risk Phrases:** Harmful by inhalation and if swallowed. Irritating to eyes, respiratory system and skin. Safety Phrase: In case of contact with eyes, rinse immediately with plenty of water and seek medical advice.

Wear suitable protective clothing.

Other hazards: No known effects from chronic exposure.

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#### 3. COMPOSITION/INFORMATION ON INGREDIENTS

Name	CAS No.	Per Cent
Calcium Flouride	7789-75-5	94-97.5
Crystalline Silica Dioxide	7631-86-9	0.09-2.8
Quartz Sand		
Calcium Carbonate	1317-65-3	1.0-2.9

#### 4. FIRST AID MEASURES

**Eye Contact** Remove contact lenses if present. Immediately rinse eyes with plenty of water,

holding eyelids open for at least 20 minutes. Consult a physician. Dust may

irritate eyes.

**Skin Contact** Remove contaminated clothing. Wash skin with water and soap. Dust: possible

skin irritation.

**Inhalation** Remove the person from exposure. Bring to fresh air. If breathing is difficult,

give oxygen. Get **immediate** medical attention. Possible irritation: mucous

membranes, upper respiratory tract and lungs.

**Ingestion** Rare in industry. Induce vomiting. Give a large quantity of water to dilute.

UNCONSCIOUS person: DO NOT induce vomiting or give any liquid. Consult

a physician.

#### 5. FIRE FIGHTING MEASURES

Flash Point Not applicable
Flammable Limits Not applicable
Auto-Ignition Temperature Not applicable

**Products of Combustion** Calcium oxide; Carbon dioxide; Hydrogen fluoride

**Fire Hazard** Dust: Flammable when exposed to heat or flames. If heated to

decomposition (1500°F) may liberate very irritating and toxic fumes or

gases (fluorine).

Flammable when exposed to flames or by chemical reaction with

oxidants.

**Explosion Hazard** Not explosive (mechanical impact). Dusts: Slightly explosive in

presence of open flames and sparks.

**Extinguishing Media** NON-FLAMMABLE. Use fire-fighting materials and procedures

adapted to the immediate environment.

**Protective Equipment** Firefighters must wear self-contained breathing apparatus (SCBA)

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Waste Disposal

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#### 6. ACCIDENTAL RELEASE MEASURES

**Spill** Avoid raising dust. Use appropriate tools to put the spilled solid in a

convenient recycling container. Finish cleaning by spreading water on the contaminated surface. Collect the dampened spilled material and place

into the recycling container.

**Personal Protection** Large concentrations of fumes or dusts: Use a self-contained breathing

apparatus (SCBA) to avoid inhalation of material.

Small concentrations: Use a NIOSH/OSHA approved full face cartridge respirator or the equivalent. Full protective clothing. Boots, Gloves. Recycle to process, if possible. Consult your local or regional authorities

for acceptable disposal methods..

#### 7. HANDLING AND STORAGE

**Handling DO NOT** ingest or inhale dusts or fumes. Keep away from incompatibles

(acids).

Ingestion or inhalation: Seek medical advice **immediately** and show the

label or the copy of this SDS to medical personnel.

**Storage** Dry, cool and well-ventilated area. Away from acids. In low fire-risk

area.

#### 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

ACGIH 2014 OSHA PEL-TWA
NIOSH REL-TWA
(mg/m³)

				(1118/111)
Name	CAS No.	PerCent	TLV-TWA	
		(%)	$(mg/m^3)$	
CalciumFluoride	7789-75-5	94-97.5	2.5 (F)	2.5 ( F)
				2.5 (F)
Silica	7631-86-9	0.09-2.8	0.025	10/%(SiO <sub>2</sub> +2) Respirable Dust
(crystalline)			(respirable	0.05
			particulate)	_
			particulate)	IDLH: 50mg/m <sup>3</sup>
Calcium	1317-65-3	1.0-2.9	Not	15 (total dust)
Carbonate			established	5 respirable fraction
				5 (respirable fraction)
				10 (Total)

ACGIH: American Conference of Governmental Industrial Hygienists.

OSHA: Occupational Safety and Health Administration.

**Note: Calcium (fluoride):** DNEL (systemic inhalation): 5 mg/m<sup>3</sup>

**Silica (amorphous):** DNEL (systemic inhalation): 4 mg/m<sup>3</sup> **Calcium (carbonate):** DNEL (systemic inhalation): 10 mg/m<sup>3</sup>

Consult local authorities for acceptable exposure limits.

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**Personal Protection** Large concentrations of fumes or dusts: Use a self-contained breathing

apparatus (SCBA) to avoid inhalation of material. Small concentrations: Use a NIOSH/OSHA approved full face cartridge respirator or the equivalent, full protective clothing, boots, gloves. Ensure that eyewash

stations and safety showers are close to the workstation location.

**Engineering Controls** Use process enclosures, local exhaust ventilation or other engineering

controls to keep airborne levels below recommended exposure limits.

#### 9. PHYSICAL AND CHEMICAL PROPERTIES

SECTION 3 – PHYSICAL AND CHEMICAL PROPERTIES

Physical State and AppearanceSolid (Crystalline powder)OdorOdorlessMolecular WeightNot applicableTasteN/ApH (1% soln/water)Not applicableColorWhite to tan

**Boiling Point** 2 500°C (4 532°F) Volatility N/A **Melting Point** 1 420°C (2 588°F) % Moisture N/A **Critical Temperature Odor Threshold** Not available N/A **Specific Gravity** Water/Oil Dist. Coeff. N/A 3.18 (Water=1) **Vapor Pressure Ionicity (in water)** Not applicable N/A **Vapor Density** Not available **Dispersion** N/A

**Solubility** No (water); Yes (ammonium salts)

#### 10. STABILITY AND REACTIVITY

**Stability** stable under normal conditions. **Conditions of Instability:** Strong acid at high temperatures

**Incompatibilities:** Reactive with acids.

**Calcium (fluoride):** with acids, chemically active metals, reducing agents, water. Contact with hot concentrated sulfuric acid: possible production of hydrofluoric acid

(Hydrogen fluoride).

Silica (crystalline): violent reaction with: fluoride, oxygen difluoride, chlorine

trifluoride.

NOTE: This list of products is not exhaustive. Verify technical documents to determine

any incompatibilities with your process.

**Corrosivity** None

#### 11. TOXICOLOGICAL INFORMATION

**Routes of Entry** Ingestion. Inhalation. Eye and skin contacts.

Acute Toxicity Calcium Fluoride: Oral acute (LD50): 4,250 mg/Kg (Rat). Intraperitoneal (LD50) >

 $1500 \text{ mg/Kg (rat)}, 2638 \text{ mg/Kg Mouse}, (RTECS)_{-}$ 

**Silica crystalline (Quartz)** Oral acute (LoTD): 120mg/Kg (rat) Inlalation acute LoTC 40mg/Kg (Mouse). Intratracheal acute (LoTD): 15.69 mg/Kg (Rat); 16.7 mg/Kg (Mouse) Intravenous acute )LoLD; 90 mg/Kg Rat); 40 mg/Kg Mouse. RTECS

Calcium Carbonate: Oral Acute LD50: 6450 (Rat). RTECS

**Irritation-sensitisation** Possibility of eyes and skin, upper respiratory tract irritation. Sensitisation.; No known

effects.

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Calcium carbonate: Severe irritation: 750 ug/24 hour (Eye, Rabbit.) Moderate

irritation: 500 mg/24 hour (Skin, Rabbit). RTECS)

Acute Effects Solid form: No health hazards. Conditions and work practices which generate dusts or

fumes should be avoided or controlled. . Ingestion and inhalation: possibility of diffuse abdominal pain, nausea, vomiting, diarrhea, thirst, saliva, albuminuria, shock.

Respiratory system. Exposure to fibrous form may cause silicosis (Fibrosis of the lungs),

inflammation, emphysema.

**Chronic Effects** No known effects from chronic exposure. Repeated or prolonged exposure

(Normal work conditions). Do not aggravate medical conditions.

Silica crystalline (Quartz): the evidence of carcinogenicity to humans is limited to crystalline silica. The silica in this product contains varying concentrations of crystalline (quartz) and amorphous silica. Target organs: respiratory system. Signs or symptoms of overexposure, shortness of breath, silicosis, emphysema..

**Calcium Carbonate:** No chronic effects of exposure have gbeen reported, Irritant for : skin eyes, nose throat respiratory tract can cause: sneezing and coughing, use as an antacid (small quantity). Calcium supplement.

**Toxicity** Workers with the following pre-existing conditions warrant particular

attention:

Silica (crystaline): tuberculosis.

Calcium (carbonate): respiratory diseases.

Eating, drinking, and smoking must be prohibited in areas where this material is handled and processed. Wash hands and face before eating, drinking, and smoking.

Calcium carbonate; Calcium fluoride: NOT A CARCINOGEN 9IARCO. NOT CLASSIFIABLE (Human, A4, ADGIH) Silica crystalline (quartz) PROVEN Group 1,

IARC) (NTP); CARCINOGEN (NIOSH SUSPECTED (Human A2, ACGIH)

CARCINOGENICITY:

MUTAGENICITY, TERATOGENICITY Calcium fluoride: Cytogenetic analysis: 1 g/Kg (Ascites tumor, Rat) (RTECS) Calcium Fluoride: INTRAPERITONEAL LoTD;: 3200 mg/Kg (9 day pregnant) (Mouse) effects on fertility: Pos-t implantation mortality (e.g. dead and/or resorbed implants per total number of implants) 67,200 mg/Kg (1-21 day pregnant) other developmental abnormalities (RTECS).

Repeated or prolonged exposure (normal work conditions): do not aggravate medical

Silica Crystalline (quartz):

DNA damages: 120mg/l/24 hour (Cells, Human; 3 mg/Kg intratraeal, Rat\_, RTECS

**Calcium** (**fluoride**): not carcinogen (IARC). Chronic overexposure: may cause increased bone density. Irritant for: skin, eyes, nose, throat and respirato (NIOSH) ry tract. May cause: coughing, chest discomfort.

**Silica (amorphous):** target organ for acute and chronic overexposure respiratory system. Possible signs after chronic overexposure: shortness of breath. Prolonged dust inhalation can cause silicosis (fibrosis of the lungs).

Calcium (carbonate): not considered a carcinogen (IARC). No chronic effects of exposure have been reported. Irritant for: skin, eyes, nose, throat, respiratory tract. Can cause: sneezing and coughing, use an antacid (small quantity); calcium supplement.

#### 12. ECOLOGICAL INFORMATION

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**Ecotoxicity** Toxic to aquatic life. May cause long terms harmful effects in

aquatic environment

**Toxicity to Animals** Calcium (fluoride): ACUTE oral (LD50): 4 250 mg/kg (rat)

Mobility (Soil) Not water soluble. Soluble compounds produced by acidic

conditions becomes mobile in water.

Biodegradation Products

Biodegradation Products (Toxicity)

BOD and COD

Not biodegradeable
Not applicable
Not available

#### 13. DISPOSAL CONSIDERATIONS

**Disposal methods** Recycle to process, if possible. Consult local or regional authorities.

If the product becomes a waste, material should be tested to determine if it must be classified as a hazardous waste under the Resource Conservation Recovery Act (RCRA 40CFR261.3). Discard in full compliance with

Federal, Provincial and local regulations.

RCRA P-Series and RCRA U-Series: Not listed.

#### 14. TRANSPORT INFORMATION

ADR Not applicable.
PIN Not applicable.
Special Provisions (Transport) Not applicable.
DOT (USA) Not Regulated

#### 15. REGULATORY INFORMATION

**TSCA** (EPA, Toxic Substance Control Act) Chemical Inventory

(40CFR710): Listed Calcium fluoride: Silica dioxide (crystalline)

CEPA DSL Canada Canadian Environmental Protection Act (DEPA) on Domestic

Substances List (DSL); Acceptable for use under the provisions of

**CEPA** 

Other Regulations CANADIAN ENVIRONMENTAL PROTECTION ACT (CEPA):

On the Domestic Substances List (DSL); Acceptable for use under

the provisions of CEPA.

Classification HCS (USA) Not regulated Classifications DSCL (EEC) Not regulated

NFPA (National Fire Protection Association) (USA)

Fire Hazard 0 Reactivity 0 Health 2 Special Hazard

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#### 16. OTHER INFORMATIO N

**References** -TLVs and BEIs (2009). Based on the Documentation of the Threshold Limit Values for Chemical Substances and Physical Agents & Biological Exposure Indices. ACGIH, Cincinnati, OH – <a href="http://www.acgih.org">http://www.acgih.org</a>

- -CCOHS (2009) Canadian Centre for Occupational Health and Safety http://www.ccohs.ca/
- -ERG (2008). Emergency Response Guidebook, US Department of Transportation, Transport Canada, et le Secretariat of Communications and Transportation of Mexico
- -HSDB (2009) Hazardous Substances Data Bank. TOXNET® Network of databases on toxicology, hazardous chemicals, and environmental health. NLM Databases & Electronic Resources, US National Library of Medicine, NHI, 8600 Rockville Pike, Bethesda, MD 20894 http://toxnet.nlm.nih.gov/cgi-bin/sis/htmlgen?HSDB
- -ESIS: C&L (Classification and Labeling), substances or preparations in accordance with Directive 67/548/EEC (substances) and 1999/45/EC (preparations),
- -ESIS: EINECS (European Inventory of Existing Commercial Chemical Substances) O.J. C 146A, 15.6.1990
- -ESIS: EINECS corrections published in O.J. C 54/13 01.03.2002, 2002/C54/08.
- -IARC Monographs on the Evaluation of Carcinogenic Risks to Humans (collection) <a href="http://www-cie.iarc.fr/">http://www-cie.iarc.fr/</a>
- -Merck Index (1999). Merck & Co., Inc., 12<sup>th</sup> edition
- -NIOSH US (2009) Pocket Guide to Chemical Hazards http://www.edc.gov/niosh/npg/
- -Patty's Industrial Hygiene and Toxicology, 3<sup>rd</sup> Revised Edition
- -Reglement sur les produits controles (Canada)
- -RTECS (2009). Registry of Toxic Effects of Chemical Substances, NIOSH, CDC
- -Toxicologie industrielle & intoxication professionnelle, 3e edition, Lauwerys
- -TSCA (2009) US EPA Toxic Substance Control Act, Chemical Inventory. System of Registries (SoR), Substance Registry Services, <a href="http://iaspub.epa.gov/sor\_internet/registry/substreg/searchandretrieve/substancesearch/search.do">http://iaspub.epa.gov/sor\_internet/registry/substreg/searchandretrieve/substancesearch/search.do</a>

Glossary ACGIH: American Conference of Governmental Industrial Hygienists.

HSDB: Hazardous Substances Data Bank.

IARC: International Agency for Research on Cancer.

NIOSH: National Institute of Occupational Safety and Health.

NTP: US National Toxicology Program.

OSHA: Occupational Safety and Health Administration.

RTECS: Registry of Toxic Effects of Chemical Substances.

#### WARRANTY

The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product.

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